

6,6'-Dimethoxy-2,2'-(hexane-1,6-diyl)dinitrilo-
dimethylidyne)diphenolH.-T. Xia,^{a*} Y.-F. Liu,^a D.-Q.
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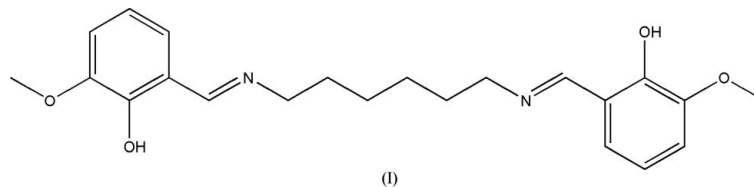
Key indicators

Single-crystal X-ray study
T = 293 K
Mean $\sigma(\text{C}-\text{C}) = 0.009 \text{ \AA}$
R factor = 0.073
wR factor = 0.277
Data-to-parameter ratio = 14.3For details of how these key indicators were
automatically derived from the article, see
<http://journals.iucr.org/e>.

The title compound, $\text{C}_{22}\text{H}_{28}\text{N}_2\text{O}_4$, is a potential multidentate azomethine ligand. The asymmetric unit contains two half-molecules, each molecule lying on a centre of symmetry. The molecules are linked into sheets and a three-dimensional network by $\text{C}-\text{H}\cdots\text{O}$ hydrogen bonds and van der Waals forces.

Comment

Schiff base salicylaldehyde derivatives have multicoordinate atoms (N and O), which can not only act as the effective organic ligand, but also exhibit a wide variety of antimicrobial activity (Key, 1974; Hodnett *et al.*, 1970; Dhar *et al.*, 1982). As part of our investigation of the reactions between 3-methoxysalicylaldehyde with diamines, we report here the crystal structure of a salicylaldehyde derivative Schiff base, (I) (Fig. 1), which was obtained by the reaction of 3-methoxysalicylaldehyde with hexane-1,6-diamine and has a *trans* configuration (*E,E*).



The asymmetric unit contains two half-molecules, each molecule lying on a centre of symmetry. The molecules are linked into a sheet parallel to the (100) plane through $\text{C}-\text{H}\cdots\text{O}$ hydrogen bonds (Fig. 2 and Table 1) and another sheet parallel to the (100) plane by van der Waals forces (Fig. 3). The alternate combination of the two (100) planes results in the formation of a three-dimensional network.

Experimental

A solution of 3-methoxysalicylaldehyde (20 mmol) and hexane-1,6-diamine (10 mmol) in ethanol (30 ml) was stirred for 8 h and then filtered. The solution was allowed to stand, yielding, by slow evaporation, yellow crystals of (I) suitable for X-ray structure analysis.

Crystal data

$\text{C}_{22}\text{H}_{28}\text{N}_2\text{O}_4$
 $M_r = 384.46$
Monoclinic, $P2_1/c$
 $a = 21.834 (11) \text{ \AA}$
 $b = 8.420 (4) \text{ \AA}$
 $c = 11.203 (6) \text{ \AA}$
 $\beta = 91.778 (9)^\circ$
 $V = 2058.6 (19) \text{ \AA}^3$

$Z = 4$
 $D_x = 1.240 \text{ Mg m}^{-3}$
Mo $K\alpha$ radiation
 $\mu = 0.09 \text{ mm}^{-1}$
 $T = 293 (2) \text{ K}$
Plate, yellow
 $0.52 \times 0.48 \times 0.15 \text{ mm}$

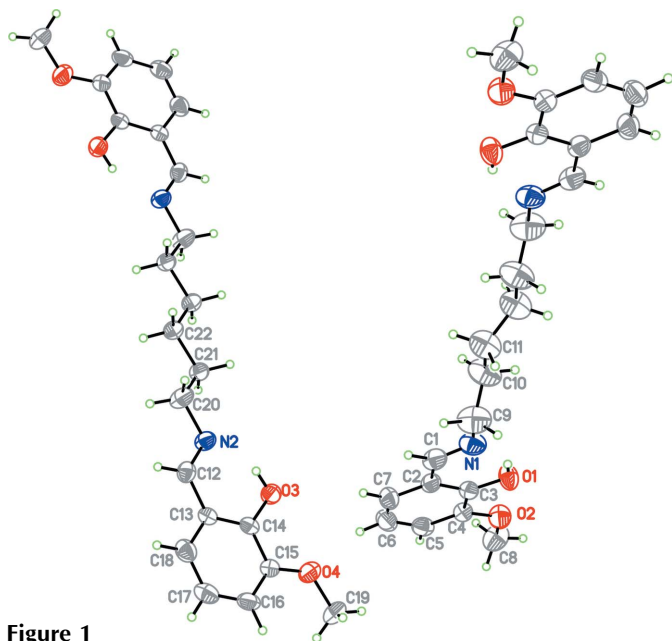


Figure 1
The structures of the two independent centrosymmetric molecules of (I), showing the atom-labelling scheme. Displacement ellipsoids are drawn at the 30% probability level. In the N1 molecule unlabelled atoms are related to labelled atoms by $(1-x, -y, 1-z)$. In the N2 molecule unlabelled atoms are related to labelled atoms by $(-x, -y, 1-z)$.

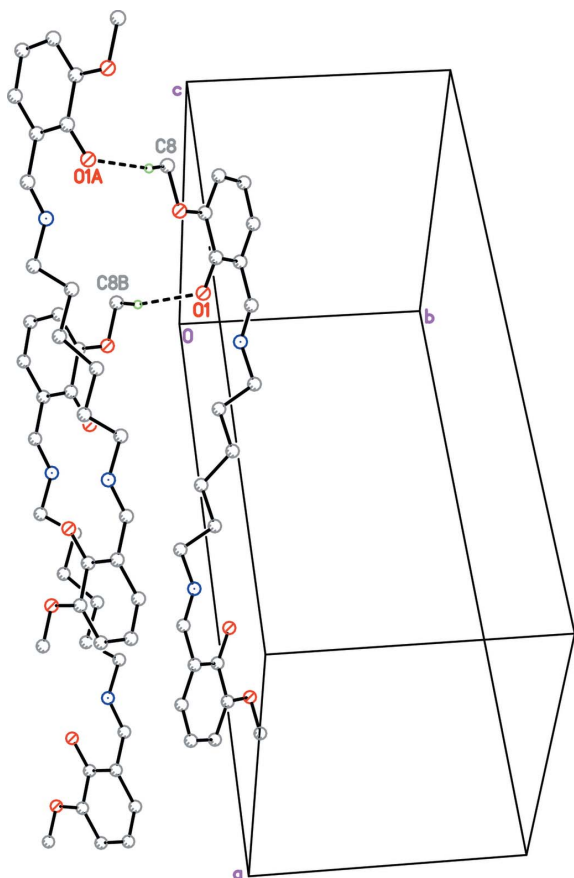


Figure 2
Part of the crystal structure of (I), showing the formation of a hydrogen-bonded sheet built from C—H...O interactions (dashed lines). For clarity, H atoms not involved in the hydrogen bonding have been omitted. [Symmetry code: (A) $x, -\frac{1}{2} - y, \frac{1}{2} + z$; (B) $x, -\frac{1}{2} - y, -\frac{1}{2} + z$].

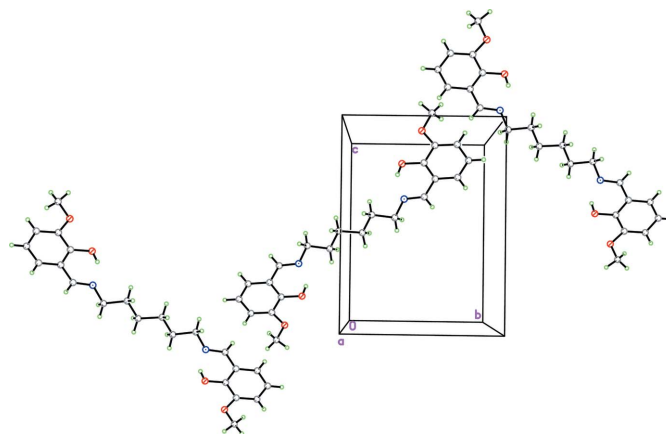


Figure 3
The crystal structure of (I), showing the formation of a sheet by van der Waals forces.

Data collection

Bruker SMART CCD area-detector diffractometer	10347 measured reflections
φ and ω scans	3621 independent reflections
Absorption correction: multi-scan (SADABS; Sheldrick, 1996)	1223 reflections with $I > 2\sigma(I)$
$T_{\min} = 0.957, T_{\max} = 0.987$	$R_{\text{int}} = 0.102$
	$\theta_{\max} = 25.0^\circ$

Refinement

Refinement on F^2	$w = 1/[\sigma^2(F_o^2) + (0.0391P)^2 + 1.1443P]$
$R[F^2 > 2\sigma(F^2)] = 0.073$	where $P = (F_o^2 + 2F_c^2)/3$
$wR(F^2) = 0.277$	$(\Delta\sigma)_{\max} < 0.001$
$S = 1.02$	$\Delta\rho_{\max} = 0.20 \text{ e } \text{\AA}^{-3}$
3621 reflections	$\Delta\rho_{\min} = -0.24 \text{ e } \text{\AA}^{-3}$
253 parameters	
H-atom parameters constrained	

Table 1
Hydrogen-bond geometry ($\text{\AA}, ^\circ$).

$D-H \cdots A$	$D-H$	$H \cdots A$	$D \cdots A$	$D-H \cdots A$
C8—H8A...O1 ¹	0.96	2.67	3.616 (8)	169

Symmetry code: (i) $x, -y - \frac{1}{2}, z + \frac{1}{2}$.

All H atoms were located in difference Fourier maps, then repositioned geometrically and treated as riding atoms, with C—H distances of 0.93 (aryl), 0.96 (methyl) and 0.82 \AA (hydroxy), and $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$ for aryl or $1.5U_{\text{eq}}(\text{C}, \text{O})$ for methyl and hydroxy groups.

Data collection: SMART (Siemens, 1996); cell refinement: SMART; data reduction: SAINT (Siemens, 1996); program(s) used to solve structure: SHELXS97 (Sheldrick, 1997a); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997a); molecular graphics: SHELXTL (Sheldrick, 1997b); software used to prepare material for publication: SHELXTL.

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